Reorientation of magnetic dipoles at the antiferroelectric-paraelectric phase transition of $Bi_{1-x}Nd_xFeO_3$ (0.15 $\leq x \leq 0.25$)

Igor Levin,^{1,*,†,§} Sarah Karimi,² Virgil Provenzano,¹ Cindi L. Dennis,¹ Hui Wu,¹ Tim P. Comyn,³ Tim J. Stevenson,³ Ronald I. Smith,⁴ and Ian M. Reaney^{2,*,‡,§}

¹National Institute of Standards and Technology, Gaithersburg, Maryland 20899, USA

²Department of Engineering Materials, University of Sheffield, Sheffield, United Kingdom

³Institute for Materials Research, University of Leeds, Leeds, United Kingdom

⁴ISIS Facility, Rutherford Appleton Laboratory, Didcot, United Kingdom

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Ceramic $Bi_{1-x}Nd_xFeO_3$ solid solutions in the approximate compositional range $0.15 \le x \le 0.25$ undergo first-order antiferroelectric (AFE) ↔ paraelectric (PE) phase transitions between 300 °C and room temperature. The orthorhombic AFE (*Pnam*, $\sqrt{2a_c} \times 2\sqrt{2a_c} \times 4a_c$, where $a_c \approx 4$ Å and is the lattice parameter of an ideal cubic perovskite) and PE (*Pbnm*, $\sqrt{2a_c} \times \sqrt{2a_c} \times 2a_c$) structures differ not only in the type of octahedral tilt system but also in the pattern of A-cation (Bi/Nd) displacements. The AFE \leftrightarrow PE transition is accompanied by a large ($\approx 2\%$) discontinuous volume change similar to that observed at the ferroelectric (FE) \leftrightarrow PE transition in BiFeO₃. For all compositions, the structural transition occurs within the G-type antiferromagnetic (AFM) state. Variable-temperature magnetic measurements reveal a strong coupling between the AFE↔PE transition and the magnetic ordering. Rietveld refinements using neutron powder-diffraction data demonstrate that the structural transition is accompanied by reorientation of magnetic dipoles within the G-type AFM array resulting in a significant increase in magnetization.

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BiFeO₃ has attracted considerable attention because of its multiferroic properties that render it a potential candidate for several device applications.^{1,2} This compound crystallizes with a perovskite-like structure and, at room temperature, exhibits rhombohedral R3c symmetry characterized by antiphase $a^{-}a^{-}a^{-}$ octahedral tilting and off-center cation displacements.^{3,4} BiFeO₃ typically cannot be synthesized as a single phase via solid-state reaction of Bi₂O₃ and Fe₂O₃. Bulk ceramic samples are commonly obtained with traces of one or both of the neighboring equilibrium phases (Bi₂FeO₉) and Bi₂₅FeO₄₀).⁵⁻⁷ On heating, the ferroelectric (FE) rhombohedral phase of BiFeO₃ undergoes a first-order phase transition at $T_{\rm C} \approx 820$ °C to a paraelectric (PE) orthorhombic GdFeO₃-type structure.⁸ This high-temperature polymorph exhibits $a^+b^-b^-$ octahedral tilting⁹ described by *Pbnm* (no. 62, nonstandard setting) symmetry and lattice parameters $\sqrt{2a_c} \times \sqrt{2a_c} \times 2a_c$ ($a_c \approx 4$ Å refers to an ideal cubic perovskite).^{8,10} The $R3c \leftrightarrow Pbnm$ transition is somewhat unusual as the symmetry is lower in the high-temperature phase and is accompanied by a large volume change (expansion on cooling) of $\approx 1.6\%$.⁸ A similar large volume change has been reported for Bi_{0.8}La_{0.2}FeO₃ (Ref. 11) and BiFe_{0.7}Mn_{0.3}O_{3+δ} (Ref. 12). For comparison, the volume change across the tetragonal-to-cubic phase transition in PbTiO₃ is $\approx 0.6\%$. BiFeO3 also undergoes paramagnetic (PM) to antiferromagnetic (AFM) ordering (on cooling) at $T_{\rm N} \approx 360 \,{}^{\circ}{\rm C}.{}^{13}$ The AFM arrangement is of the G type modified by a longwavelength (≈ 620 Å) spiral modulation associated with a canting of magnetic moments.¹³ This modulation allegedly inhibits development of a net magnetization in BiFeO₃. A-site substitutions of rare-earth (RE) cations (La, Sm, Gd, and Nd) in BiFeO₃ suppress the spiral magnetic modulation yielding a detectable ferromagnetic response.^{14–16}

In the solid solution Bi_{1-x}Nd_xFeO₃, an antiferroelectric

(AFE) PbZrO₃-type orthorhombic phase exists between 300 °C and room temperature for compositions $0.15 \le x \le 0.25$ ¹⁷ Similar results have also been reported for epitaxial thin films of (Bi_{0.86}Sm_{0.14})FeO₃.¹⁸ The PbZrO₃ structure (*Pbam*, $\sqrt{2a_c \times 2}\sqrt{2a_c \times 2a_c}$) combines $a^0b^-b^$ octahedral tilting with antipolar Pb displacements¹⁹ (Fig. 1, denoted as O_2). X-ray powder-diffraction patterns of AFE (Bi, Nd)FeO₃ can be fully indexed assuming a PbZrO₃-type structure; however, electron diffraction revealed additional superstructure reflections which require a $\sqrt{2a_c} \times 2\sqrt{2a_c} \times 4a_c$ supercell.¹⁷ These extra reflections were attributed to a complex NaNbO₃-like $(a^-a^-c^+)/(a^-a^-c^-)$ octahedral tilting²⁰ but the exact structure of the antipolar (Bi,Nd)FeO₃ phase remained uncertain.¹⁷ Tentative phase

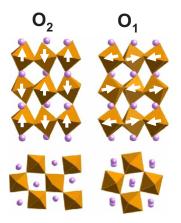


FIG. 1. (Color online) Schematic rendering of the structures for the O_1 and O_2 polymorphs in Bi_{0.825}Nd_{0.175}FeO₃ (x=0.175) refined using neutron-diffraction data collected at 25 °C and 350 °C, respectively. Directions of the magnetic moments are indicated using white arrows.

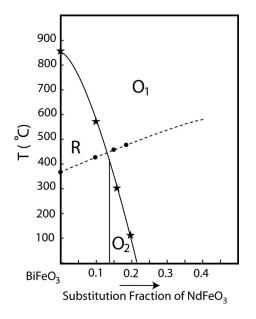
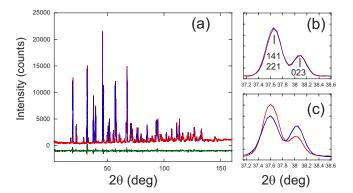


FIG. 2. A T_0 diagram for the Bi-rich part of the BiFeO₃-NdFeO₃ system. Solid and dashed lines correspond to the structural and magnetic phase transitions, respectively. Experimentally determined transition temperatures (present study) are indicated using symbols. T_0 values for the magnetic transition were inferred from the variable-temperature magnetic measurements. **R** refers to the rhombohedral structure, whereas **O**₁ and **O**₂ refer to the GdFeO₃- and PbZrO₃-type structures, respectively.

diagrams for the (Bi, RE)FeO₃ systems²¹ suggested the phase transition sequence R3c (labeled as **R**) \leftrightarrow Pbnm(**O**₁) and Pbam(**O**₂) \leftrightarrow Pbnm(**O**₁), depending on composition (Fig. 2). In the present study, we combined high-temperature x-ray, neutron, and electron diffraction to further study the transitions in ceramic Bi_{1-x}Nd_xFeO₃ (0 $\leq x \leq 0.2$), a detailed account of which will be presented elsewhere. This contribution, however, focuses on the **O**₂ \leftrightarrow **O**₁ (AFE \leftrightarrow PE) phase transition and its effect on the magnetic structure and properties.

Bi_{1-x}Nd_xFeO₃ samples (x=0.1, 0.15, 0.175, and 0.2) were prepared by conventional solid-state synthesis. Bi₂O₃ (99.999%), Nd₂O₃ (99.999%), and Fe₂O₃ (A.G.) powders were mixed in acetone using an agate mortar and pestle, dried, pressed into pellets and heat treated in air at 850 °C for 5 h. The Nd₂O₃ powder was preheated at 700 °C prior to weighing to remove any residual moisture. The samples were subjected to multiple heat treatments (with intermediate grindings) at 920 °C for a total time of 200 h until no detectable changes in the width of x-ray reflections could be observed. Thermogravimetric analyses indicated no detectable weight loss under these heat-treatment conditions.

X-ray diffraction (XRD) patterns were collected using a Panalytical X'pert Pro diffractometer equipped with a Ge single-crystal incident beam monochromator (Cu $K_{\alpha l}$ radiation), a Pixcel position sensitive detector, and an Anton Paar HTK-16 high-temperature stage. A Philips CM30 TEM operated at 200 kV and equipped with a double-tilt high-temperature stage was used for electron diffraction and imaging. Neutron powder-diffraction data were collected using the BT-1 [wavelength 1.5402(2) Å] and POLARIS (time-of-



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FIG. 3. (Color online) (a) Upper: experimental (red/symbols) and calculated (blue/line) constant-wavelength neutron-diffraction profiles for the O_2 phase of Bi_{0.825}Nd_{0.175}FeO₃ (x=0.175) at room temperature. Lower: residual (R_{wp} =4.18%, χ^2 =2.44). The calculated profile corresponds to the PbZrO₃-like model with *Pbam* symmetry; (b) fit for a pair of reflections (141/221 and 023) for the models with *c*- (b) and *a*-axis (c) orientations of the magnetic moments. Similar results were obtained for x=0.15.

flight) instruments located at NIST and ISIS (Rutherford Appleton Laboratory), respectively. In both cases, the sample powders were loaded in vanadium containers. Rietveld refinements were performed using the GSAS software package.²² The SARAh software²³ was combined with GSAS for symmetry-constrained magnetic-structure refinements. Variable field (from $+5570 \text{ kA/m} = 70\ 000 \text{ Oe}$ to -5570 kA/m) and temperature (from 300 to 1000 K) magnetic measurements were performed on sintered pellets using a Quantum Design superconducting quantum interference device vibrating-sample magnetometer equipped with a sample oven that extends the maximum operating temperature to 1000 K. Before each variable-temperature scan, the sample was first subjected to a strong magnetic field of 70 000 Oe, and the field was then reduced to a selected value.

As described previously,17 room-temperature XRD patterns for x=0.1 are accounted for by the rhombohedral R3cstructure, whereas those for $0.15 \le x \le 0.2$ can be fitted using a PbZrO₃-type Pbam structure. The XRD patterns reveal no clear evidence for a doubling of the unit cell from $\sqrt{2a_c \times 2} \sqrt{2a_c \times 2a_c}$ to $\sqrt{2a_c \times 2} \sqrt{2a_c \times 4a_c}$, as suggested by the observation of $\frac{1}{4}[001]_c$ -type superlattice reflections in electron diffraction.¹⁷ Likewise, room-temperature neutron powder-diffraction data collected for x=0.15 and x=0.175are satisfactorily fitted using the PbZrO₃-like Pbam model which readily accounts for all observed reflections (Fig. 3). Nevertheless, large and highly anisotropic atomic displacement parameters for the oxygen and Bi/Nd positions indicate that the actual structure for these samples deviates slightly from *Pbam* symmetry. Reflection conditions 0kl:k+l=2nand h0l:h=2n in electron-diffraction patterns for the samples with $0.15 \le x \le 0.2$ (Ref. 17) limit the choice of possible space groups for the $\sqrt{2a_c} \times 2\sqrt{2a_c} \times 4a_c$ superstructure to Pnam (no. 62) and Pna21 (no. 33). Pnam symmetry matches a structure with NaNbO3-like octahedral tilting and PbZrO₃-like antipolar displacements of the A cations; such complex $(a^{-}a^{-}c^{+})/(a^{-}a^{-}c^{-})$ tilting appears plausible considREORIENTATION OF MAGNETIC DIPOLES AT THE ...

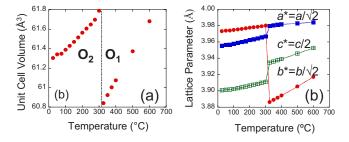


FIG. 4. (Color online) Temperature dependence of the unit-cell volume (a) and lattice parameters (b) for x=0.15. The unit-cell volume and lattice parameters were reduced to those of a cubic perovskite cell with $a_c \approx 4$ Å. In Fig. 1, the *a* axis is parallel to the viewing direction and the *b* axis is horizontal for both O₁ and O₂.

ering a competition between the $a^-a^-a^-$ and $a^+b^-b^-$ tilts exhibited by the parent phases BiFeO₃ and NdFeO₃, respectively. Attempts to refine this model using the neutron-diffraction data produced no significant improvement in the fit, though relatively strong correlations among structural variables in the larger model precluded refinements of anisotropic displacement parameters. We conclude that NaNbO₃-like tilting is not strong and/or sufficiently ordered to significantly contribute to the powder-diffraction patterns at room temperature.

High-temperature x-ray, electron-, and neutron-diffraction experiments confirmed that all (Bi,Nd)FeO₃ samples undergo a phase transition on heating to the O₁ GdFeO₃-like structure (Fig. 1) in agreement with the proposed phase diagram (Fig. 2). The transition temperature decreases rapidly from $\approx 600 \,^{\circ}$ C for x=0.10 to $\approx 300 \,^{\circ}$ C for x=0.15 to $\approx 120 \,^{\circ}$ C for x=0.20 (Fig. 2).²⁴ Despite the same no. 62 space group for both the O₂ and O₁ structures, their symmetries do not obey group-subgroup relations and the O₂ \leftrightarrow O₁ transition is first order. Likewise, the $\mathbf{R} \leftrightarrow \mathbf{O}_2$ transition (Fig. 2), which presumably occurs in a very narrow compositional range, must be first order. A large discontinuous volume change, similar to that encountered in BiFeO₃, is observed

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upon both $\mathbf{R} \leftrightarrow \mathbf{O}_1$ and $\mathbf{O}_2 \leftrightarrow \mathbf{O}_1$ transitions [Fig. 4(a)], even though the structural changes underlying these two transitions are different. This large volume change at relatively low temperatures explains the fragility of the sintered pellets after cooling to room temperature. Figure 4(b) summarizes changes in the lattice parameters (reduced to a pseudocubic ≈ 4 Å perovskite cell) across the $\mathbf{O}_2 \leftrightarrow \mathbf{O}_1$ transition for x=0.15. A large volume effect is determined by the changes in the b^* dimension (horizontal in Fig. 1). The exact structural changes/distortions underlying the large change in the *b*-axis parameter remain to be clarified. Detailed neutron *total* scattering studies aimed at simultaneous analyses of changes in the local and average distortions across the $\mathbf{O}_2 \leftrightarrow \mathbf{O}_1$ phase transition are planned and will be reported separately.

Rietveld refinements of the room-temperature magnetic structure for all compositions confirm an overall G-type AFM ordering (Fig. 3) but powder-diffraction data do not support refinements of a spiral modulation, if present. In the O_2 phase, the magnetic moments are aligned with the *c* axis of the $\sqrt{2a_c} \times 2\sqrt{2a_c} \times 2a_c$ cell (Fig. 1, left) as evidenced by the characteristic intensity distribution for a pair of reflections 141/221 and 023 in the neutron-diffraction patterns (Fig. 3). The refined value of the magnetic moment for Fe^{3+} is $m=3.75(2)\mu_{\rm B}$. The directions of magnetic moments in the O_2 phase are orthogonal to those in the room-temperature AFM Pbnm structure of NdFeO₃. The AFM ordering is preserved upon the $O_2 \leftrightarrow O_1$ transition but the magnetic moments in the high-temperature O_1 phase become predominantly aligned with the *a* axis ($\approx \sqrt{2a_c}$) (Fig. 1, right); that is, neutron diffraction suggests a pronounced coupling between the displacive and magnetic ordering manifested in the reorientation of magnetic moments by 90° at the displacive phase transition (Fig. 1). NdFeO₃ also undergoes reorientation of magnetic moments (without a structural transition) but in the reverse direction, i.e., from aligned parallel to the *a* axis at room temperature to the c axis below 190 K.²⁵

Magnetic measurements provide further evidence for a significant effect of the displacive transition $O_2 \leftrightarrow O_1$ on the

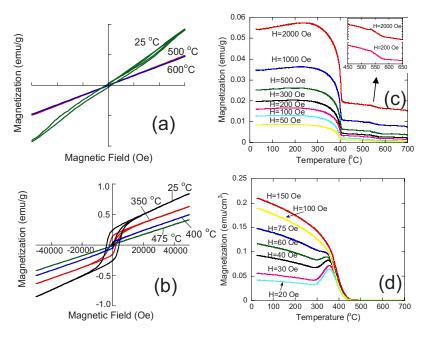


FIG. 5. (Color online) Field (*H*) dependence of magnetization (*M*) for the (a) $Bi_{0.9}Nd_{0.1}FeO_3$ and (b) $Bi_{0.85}Nd_{0.15}FeO_3$ samples at selected temperatures. [(c) and (d)] Temperature dependence of magnetization under different external fields for the same samples, respectively. Prior to each variable-temperature scan the samples were first subjected to a magnetic field of 70 000 Oe, and, then, the field was reduced to a selected value.

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cilitated by applied magnetic field. For x=0.2, where the

structural transition occurs well below the temperature of

magnetic ordering, the peak of magnetization was evident

change encountered during the high-temperature phase tran-

sition in BiFeO₃ is maintained upon Nd substitution and is

particularly evident for the AFE \leftrightarrow PE transition between the

PbZrO₃-like and GdFeO₃-like structures (Fig. 2). Magnetic

measurements confirm a robust FM response for the AFE

phase and reveal a strong effect of the structural phase tran-

sition on magnetization. According to neutron diffraction, the

AFE \leftrightarrow PE transition is accompanied by a $\approx 90^{\circ}$ reorientation

of the magnetic dipoles in the G-type AFM array. More de-

tailed studies of the interplay between the structural and

magnetic transitions in this and other rare-earth doped

BiFeO₃ systems are warranted to understand changes in

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magnetization with field and temperature.

In summary, our results demonstrate that a large volume

even at large (e.g., 2000 Oe) fields.

magnetic response (Fig. 5). At room temperature, the x=0.1 sample exhibits only a slim hysteresis loop [Fig. 5(a)], whereas the x=0.15 [Fig. 5(b)] and x=0.20 (not shown) samples yield robust loops with coercive fields of $H_c \approx 4170$ Oe and $H_c \approx 1950$ Oe, respectively; in all cases, the response changes to paramagnetic at high temperatures. The magnetic phase transition is clearly manifested in a large increase in magnetization [Figs. 5(c) and 5(d)]. The origin of the ferromagnetic response in the Nd-doped BiFeO₂ requires further investigation. Importantly, a pronounced peak of magnetization was also observed around the temperatures of the structural phase transition; the temperature of peak magnetization varies somewhat with the applied field. For x=0.1, where the structural transition occurs in the PM state, the peak is weak [inset in Fig. 5(c)] and most prominent for larger fields. In contrast, for x=0.15 [Fig. 5(d)], this peak becomes pronounced for weak fields because for strong fields the effect is masked by a marked increase in magnetization upon crossing the magnetic transition. The results concur with the reorientation of magnetic dipoles upon the structural transition as suggested by neutron refinements. Conceivably, even in the PM state, the magnetic moments are preferentially aligned with one of the crystallographic axes, and the weak peak of magnetization for x=0.1 is caused by reorientation of the moments which is further fa-

*Corresponding author.

- [†]igor.levin@nist.gov
- [‡]i.m.reaney@sheffield.ac.uk
- [§]Also at Department of Materials Science and Engineering, University of Maryland, College Park, MD 20742, USA.
- ¹G. Catalan and J. F. Scott, Adv. Mater. **21**, 2463 (2009).
- ²R. Ramesh and N. A. Spaldin, Nature Mater. 6, 21 (2007).
- ³J. M. Moreau, C. Michel, R. Gerson, and W. J. James, J. Phys. Chem. Solids **32**, 1315 (1971).
- ⁴F. Kubel and H. Schmid, Acta Crystallogr. B 46, 698 (1990).
- ⁵R. Palai, R. S. Katiyar, H. Schmid, P. Tissot, S. J. Clark, J. Robertson, S. A. T. Redfern, G. Catalan, and J. F. Scott, Phys. Rev. B **77**, 014110 (2008).
- ⁶M. Valant, A.-K. Axelsson, and N. Alford, Chem. Mater. **19**, 5431 (2007).
- ⁷ M. W. Lufaso, T. A. Vanderah, I. M. Pazos, I. Levin, R. S. Roth, J. C. Nino, V. Provenzano, and P. K. Schenck, J. Solid State Chem. **179**, 3900 (2006).
- ⁸D. C. Arnold, K. S. Knight, F. D. Morrison, and P. Lightfoot, Phys. Rev. Lett. **102**, 027602 (2009).
- ⁹A. M. Glazer, Acta Crystallogr. B 28, 3384 (1972).
- ¹⁰M. Marezio, J. P. Remeika, and P. D. Dernier, Acta Crystallogr.
 B 26, 2008 (1970).
- ¹¹M. Polomska, W. Kaczmarek, and Z. Pajak, Phys. Status Solidi 23, 567 (1974).
- ¹²S. M. Selbach, Th. Tybell, M.-A. Einarsrud, and T. Grande, Phys. Rev. B **79**, 214113 (2009).
- ¹³I. Sosnowska, T. Peterlin-Neumaier, and E. Steichele, J. Phys. C

15, 4835 (1982).

sions.

- ¹⁴G. L. Yuan, S. W. Or, J. M. Liu, and Z. G. Liu, Appl. Phys. Lett. 89, 052905 (2006).
- ¹⁵ V. A. Khomchenko, V. V. Shvartsman, P. Borisov, W. Kleeman, D. A. Kiselev, I. K. Bdikin, J. M. Vieira, and A. L. Kholkin, Acta Mater. **57**, 5137 (2009).
- ¹⁶K. S. Nalwa and A. Garg, J. Appl. Phys. **103**, 044101 (2008).
- ¹⁷S. Karimi, I. M. Reaney, I. Levin, and I. Sterianou, Appl. Phys. Lett. **94**, 112903 (2009).
- ¹⁸C. J. Cheng, D. Kan, S. H. Lim, W. R. McKenzie, P. R. Munroe, L. G. Salamanca-Riba, R. L. Withers, I. Takeuchi, and V. Nagarajan, Phys. Rev. B **80**, 014109 (2009).
- ¹⁹E. Sawaguchi, H. Maniwa, and S. Hoshino, Phys. Rev. 83, 1078 (1951).
- ²⁰A. M. Glazer, M. Ahtee, and H. D. Megaw, Acta Crystallogr. A 28, S179 (1972).
- ²¹S. Karimi, I. M. Reaney, Y. Han, J. Pokorny, and I. Sterianou, J. Mater. Sci. 44, 5102 (2009).
- ²²A. C. Larson and R. B. Von Dreele, Los Alamos National Laboratory LAUR Report No. 86-748, 1994 (unpublished).
- ²³A. S. Wills, Physica B **276**, 680 (2000).
- ²⁴ The transition temperatures reported in this study should be considered as approximate. Accurate determination of these temperatures is limited by the difficulties in achieving a fully homogeneous distribution of Bi and Nd using solid-state synthesis.
- ²⁵J. Bartolomé, E. Palacios, M. D. Kuz'min, F. Bartolome, I. Sosnowska, R. Przenioslo, R. Sonntag, and M. M. Lukina, Phys. Rev. B 55, 11432 (1997).